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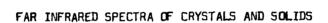
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ANNUAL SUMMARY REPORT .



IN A LARGE RANGE OF TEMPERATURES.

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#### A - ABSTRACT.

Far Infrared spectra of 11 solids at liquid helium temperature and 7 at room temperature.

#### B - TABLE OF CONTENTS.

#### I - WORK EXECUTED DURING THE FIRST TWO YEARS PERIOD (1/5/1961 to 30/4/1963).

- 1. Instrumentation. -
- 2. Far Infrared absorption spectra at liquid helium temperature. -
- 3. Far Infrared absorption spectra at room temperature. -
- 4. Theory and applications. -

#### II - PRINCIPLE RESULTS. -

III - BRIEF TECHNICAL DESCRIPTION OF THE WORK CONTEMPLATED ON FAR INFRARED SPECTRA OF

CRYSTALS AND SOLIDS IN A LARGE MANGE OF TEMPERATURES DURING THE PERIOD OF EXTENSION (ONE YEAR).-

#### IV - REFERENCES. -

C -- BODY OF THE REPORT : FAR INFRARED SPECTRA OF CRYSTALS AND SOLIDS IN A LARGE

OF TEMPERATURES. --

D - INDEX CARDS. -

This is a summary of the work described in administrative reports no 1 to 7, with some additives covering the period from 31/1/1963 to 30/4/1963.

I - WORK EXECUTED DURING THE FIRST TWO YEARS PERIOD . (1/3/1961 to 30/4/ 1963). -

#### 1. - Instrumentation. -

- a) Development of Far Infrared techniques at low temperature: the helium is recycled and we established the connexion of our laboratory to an outside gazometer at the end of December 1961. Our first liquid helium cryostat was ready since October 1961 (Fig. 18). There has been some difficulties to locate accurately the cryostat inside the evacuated spectrometer (Fig. 1).
- b) Development of a polariser (<u>Fig. 2</u>): we use a pile of 20 sheet of polyethylene (thickness 30 microns).

#### 2. - Far Infrared absorption spectra at liquid helium temperature. -

We got the first spectra at liquid helium temperature in May 1962 and during the last 10 months we studied:

Ge ( $\underline{\text{Fig. 3 - 4}}$ ), Quartz ( $\underline{\text{Fig. 5}}$ ), Silica ( $\underline{\text{Fig. 21}}$ ), I Cs ( $\underline{\text{Fig. 9}}$ ) Al<sub>2</sub>0<sub>3</sub>, F<sub>2</sub>Ca ( $\underline{\text{Fig. 12}}$ ), F<sub>2</sub>Sr ( $\underline{\text{Fig. 12}}$ ), F<sub>3</sub>Tm ( $\underline{\text{Fig. 13}}$ ), Cu<sub>2</sub>O ( $\underline{\text{Fig. 15}}$ ), Te Hg.

Except for Germanium, this seems to be the first collection of Far Infrared data at liquid helium temperature.

#### 3. - Fer Infrared absorption spectra at room temperature. -

At room temperature we studied SnI<sub>4</sub> (<u>Fig. 14</u>), Cu<sub>2</sub>O (<u>Fig. 15 - 16</u>), Gypsum, (CN)<sub>2</sub>Hg, Cl<sub>2</sub>Hg, Cl<sub>2</sub>Hg<sub>2</sub>, NaCl and all the above mentioned crystals. Calcite (CO<sub>3</sub> Ca) (<u>Fig. 19</u>) and Siderose (CO<sub>3</sub>Fe) (<u>Fig. 20</u>) have been studied with pola in rised radiations.

#### 4. - Theory and Applications. -

a) The need for a large collection of well established accurate infrared spectra of solids has been stressed by some scientists (ref. 1 for instance). in

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order to develop a theory of lattice dy amics which would lie on solid experimental basis. It is too soon to try to relate completely our spectra to lattice dynamics.

- b) In a first stade we try a LORENTZ one :  $\beta$  (Strength of the resonance),  $\delta$  width,  $\lambda_n$  (wavelength),  $\xi_0$  (static dielectric constant).
- c) The general behaviour of the Far Infrared wing of the absorption band of single crystals at low temperature has a simple interpretation in quantum theory.
- d) The absorption coefficient of Quartz (single crystal) (<u>Fig. 5</u>) and molten quartz (glass) (<u>Fig. 21</u>) differ remarkably and their behaviour with temperature seams strikingly different.

#### II - PRINCIPLE RESULTS. -

1. - Normal frequencies have been ascribed for :

I Cs (2):  $\lambda_0$  = 161 microns; Cu<sub>2</sub>O (5): we cannot agree completely with PASTER-NIAK's assignment and we propose for the 2 Infrared active vibrations:  $\lambda_1$  = 16,4 microns;  $\lambda_2$  = 60 microns; for  $\lambda_4$  (optically inactive) and  $\lambda_3$  (Raman active), we accept the assignments of PASTERNIAK:  $\lambda_4$  = 8,9 microns;  $\lambda_3$  = 55 microns.

 $SnI_{A}$  (3); (CN) Hg (5); CL Hg (5).

- 2. Dichroïsm of the 70 microns absorption band of quartz (6). YAROSLAWSKI's suggestion of hydrogen bending of impurities ought to be considered.
- 3. Effect of temperature on the Far Infrared lattice absorption spectra of single crystals (2-6-7).
  - a) The maximum of absorption is shifted to shorter wavelengths when the temperature is decreased (see quartz, Cu<sub>2</sub>O, ICs).
  - b) The transmission increases and this effect is highly more important on the low frequency wing of the absorption band (see Quartz, Cu<sub>2</sub>O, ICs, Thulium fluoride).

The effect a) was generally awaited: the importance of effect b) was unspected and our laboratory has probably been the first to establish it on a large number of single crystals. It is interesting with in two points of views:

- theoretical : it seems to confirm that vertical transitions from acoustical to optical branches of any wave vector are infrared active in a second order approximation. In other words any absorption on the low frequency wing of Far Infrared bands may be explained as a combination of one infrared photon with one acoustical phonon, to give one optical phonon. This absorption disappears at low temperatures when there is a lack of phonons in the crystal.
- technical: if it is true that at room temperature there is a great lack of materials transparent in the Far Infrared, the situation is reversed at low temperatures where a lot of single crystals get completely transparent. We have shown the possibility to select transparent hosts at liquid helium temperature where ions to be studied in the Far Infrared could be included (7) (Fig 22). Neodymium nitrate and Semarium nitrate have a lot of common lines ascribed to lattice vibrations and one specific line at 300 microns, ascribed to Nd<sup>3+</sup>.

## III - BRIEF TECHNICAL DESCRIPTION OF THE WORK CONTEMPLATED ON FAR INFRARED SPECTRA OF CRYSTALS AND SOLIDS IN A LARGE RANGE OF TEMPERATURE DURING THE PERIOD OF EXTENSION (one year).

crystals) at

- 1. Far Infrared spectra of CuCl CuBr CuI NaCl KBr (single crystals) at various temperatures.
- 2. idem for Ag<sub>2</sub>O compared to Cu<sub>2</sub>O .
- 3. Idem for La Br<sub>3</sub>, laCl<sub>3</sub>, Nd Cl<sub>3</sub>, Pr Cl<sub>3</sub> (single crystals grown by Prof. Boris STOICHEFF).
- 4. We shall have to check again the spectra of silica and other glasses (E203)at low temperatures. It would be good to put the plate of glass between 2 plates of quartz when thermal conductivity is much higher. Up to now we have not found any difference between single crystals and pellets of powders. We shall try to get smaller particles and check again the spectra in order to find an explanation. It will be worthwhile to compare the spectra of Irtran and of a single crystal of S Zn.
- 5. Spectra of some molecular crystals (for instance Cl<sub>2</sub>Hg<sub>2</sub>, (CN)<sub>2</sub>Hg, naphtalene etc....) at liquid helium temperature, where, up to now, there is not any experimental data.

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#### IV - REFERENCES. -

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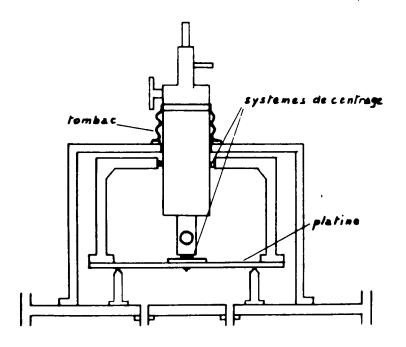


Fig. 1

Position du cryostat dans le spectroscope

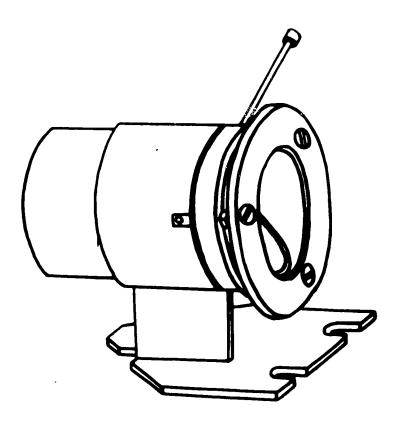
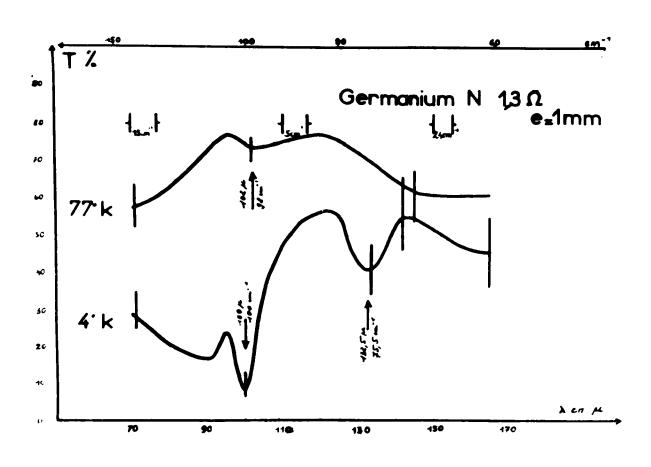
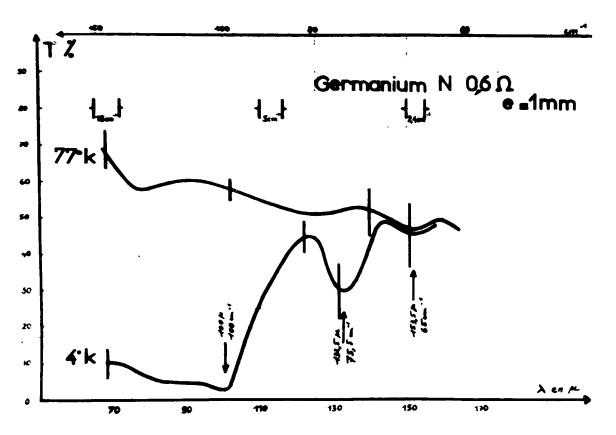


Fig. 2

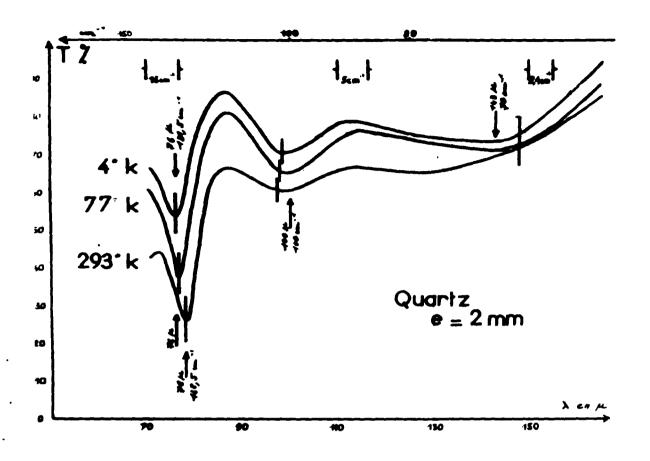
Croquis du polariseur



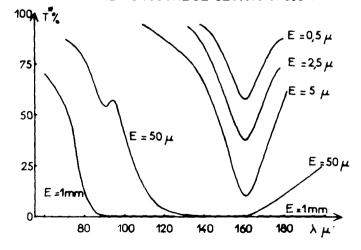


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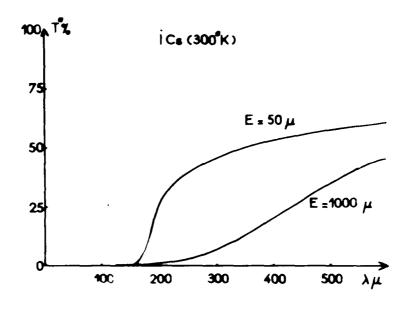


Fig. 7

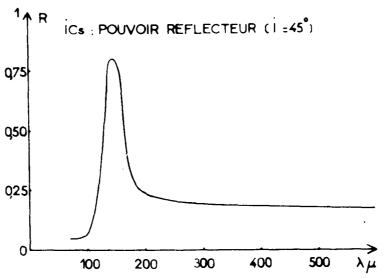


Fig. 8



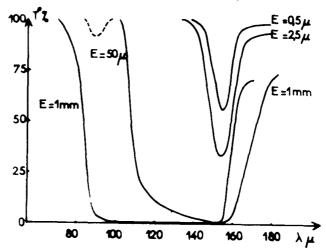


Fig. 9

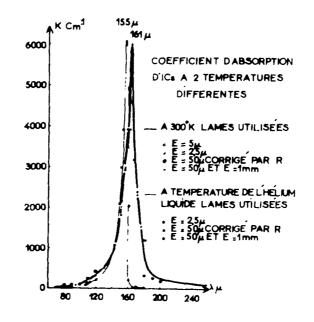


Fig. 10

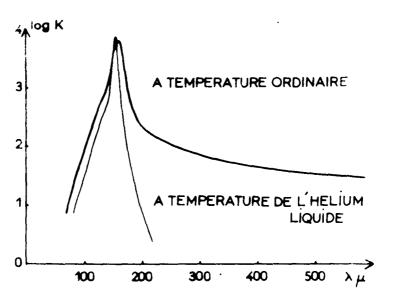


Fig. 11

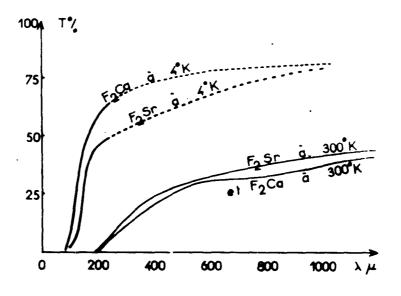
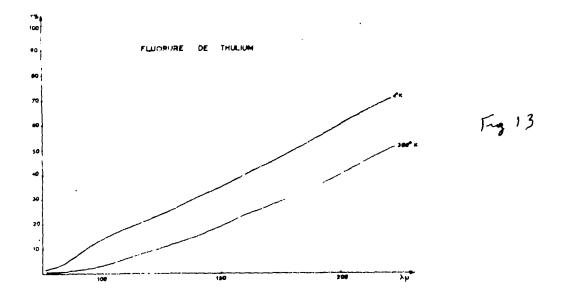
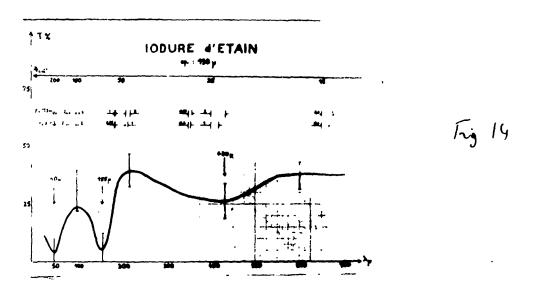


Fig. 12





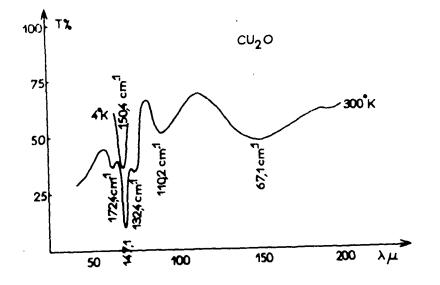


Fig. 15.

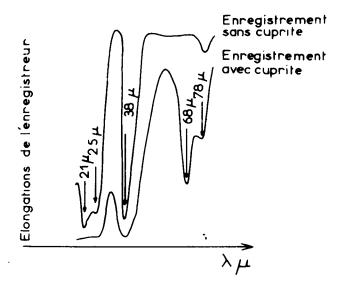


Fig. 16

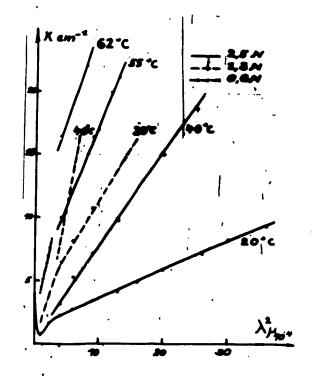


Fig 17

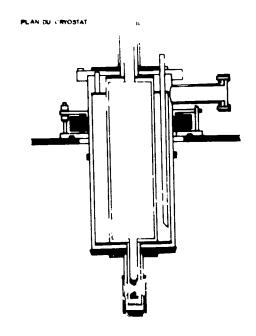
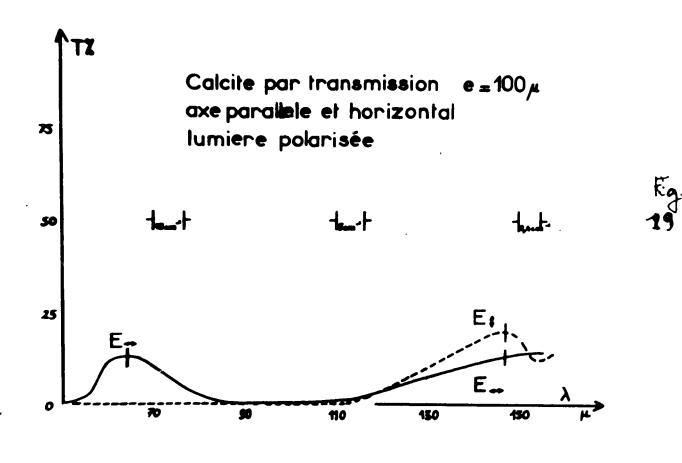
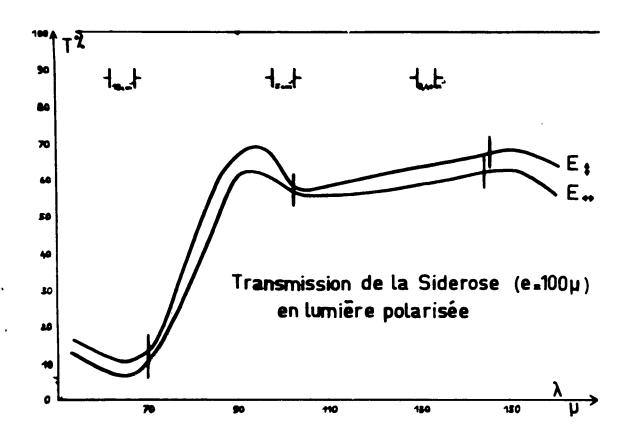
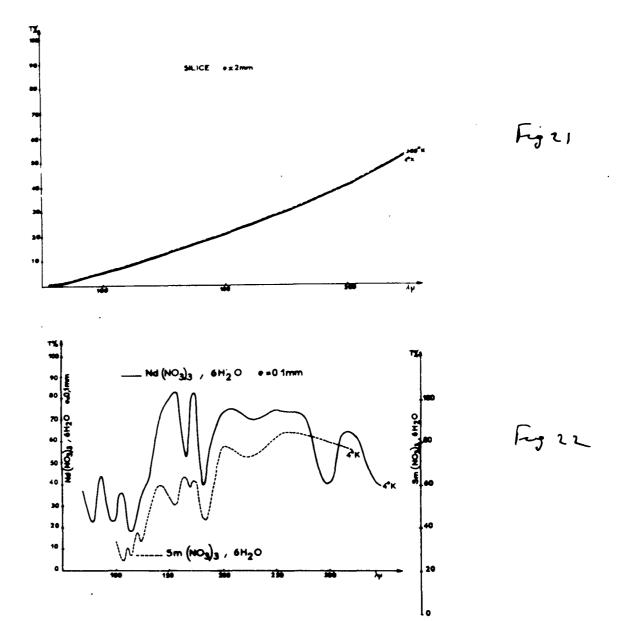


Fig 18







GROUP FOR THE ADVANCEMENT OF SPECTROSCOPIC METHODS 1, rue Gaston-Boissier PARIS / 15ème (FRANCE) 31 May 1963 ANNUAL SUMMANT REPORT	GROUP FOR THE ADVANCEMENT OF SPECTROSCOPIC METHODS 1, rue Gaston-Bolssier PARIS / 15ème (FRANCE) 31 May 1963 ANNUAL SUMMARY REPORT
FAR INFRARED SPECTRA OF CRYSTALS AND SOLIDS IN A LARGE RANGE OF TEMPERATURE.  J. LECOMIE and A. HADNI.  ABSTRACT: Far Infrared spectra of 11 solids at liquid helium temperature and 7 at room temperature.	FAR INFRARED SPECTRA OF CRYSTALS AND SOLIDS IN A LARGE RANGE OF TEMPERATURE.  J. LECOMTE and A. HADNI.  ABSTRACT: Far Infrared spectra of 11 solids at liquid helium temperature and 7 at room temperature.
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